

# Equilibrium Reactor

Spoken Tutorial Project  
<http://spoken-tutorial.org>

National Mission on Education through ICT  
<http://sakshat.ac.in>

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# Learning Objectives

**In this tutorial, we will learn to:**



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In this tutorial, we will learn to:

- Define an **Equilibrium Reaction**



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- Define an **Equilibrium Reaction**
- Simulate an **Equilibrium Reactor**



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In this tutorial, we will learn to:

- Define an **Equilibrium Reaction**
- Simulate an **Equilibrium Reactor**
- Calculate **Conversion percentage** and **Reaction extent**



# System Requirement



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- **DWSIM v 5.2 (Classic UI) Update 22**



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- Windows 10





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- DWSIM v 5.2 (Classic UI) Update 22
- Windows 10
- Any OS: Linux, Mac OS X or FOSSEE OS on ARM



# Prerequisites

To practice this tutorial, you should know



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To practice this tutorial, you should know

- Add components to a **flowsheet**



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To practice this tutorial, you should know

- Add components to a **flowsheet**
- Select **thermodynamic** packages



# Prerequisites

To practice this tutorial, you should know

- Add components to a **flowsheet**
- Select **thermodynamic** packages
- Add **material** and **energy** streams and specify their properties



# Prerequisite Tutorials and Files

- <http://spoken-tutorial.org>
- You can access these tutorials and all the associated files from this site



# Reaction and Inlet Condition

<b>Reaction</b>	<b><math>\text{CO(g)} + \text{H}_2\text{O(g)} \rightleftharpoons \text{H}_2\text{(g)} + \text{CO}_2\text{(g)}</math></b>	
<b>Inlet Stream</b>	<b>Mass Flow</b>	<b>3600 kg/h</b>
	<b>Temperature</b>	<b>25 °C</b>
	<b>Pressure</b>	<b>1.01325 bar</b>
	<b>Mole Fraction</b>	<b><math>x_{\text{CO}} = 0.5</math></b>
		<b><math>x_{\text{H}_2\text{O}} = 0.5</math></b>
		<b><math>x_{\text{H}_2} = 0</math></b>
		<b><math>x_{\text{CO}_2} = 0</math></b>



# Property Package and Reaction Temperature

- Property Package:  
Raoult's Law
- Reaction Temperature:  
225 °C





# Summary

In this tutorial, we have learnt to:

- Define an Equilibrium Reaction
- Simulate an Equilibrium Reactor
- Calculate Conversion percentage and Reaction extent



# Assignment

Reaction	$\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$	
Package	Peng-Robinson	
Inlet Stream	Mass Flow Temperature Pressure Mole fraction	1000 kg/h 425 °C 200 bar $x_{\text{N}_2} = 0.5$ $x_{\text{H}_2} = 0.5$ $x_{\text{NH}_3} = 0$
Reaction Temp	400 °C	



# About the Spoken Tutorial Project

- Watch the video available at [http://spoken-tutorial.org/What\\_is\\_a\\_Spoken\\_Tutorial](http://spoken-tutorial.org/What_is_a_Spoken_Tutorial)
- It summarises the Spoken Tutorial project



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- It summarises the Spoken Tutorial project
- If you do not have good bandwidth, you can download and watch it



# Spoken Tutorial Workshops

## The Spoken Tutorial Project Team,

- Conducts workshops using spoken tutorials
- Gives certificates to those who pass an online test
- For more details, please write to [contact@spoken-tutorial.org](mailto:contact@spoken-tutorial.org)



# Forum for specific questions

- Do you have questions in this Spoken Tutorial?
- Please visit <http://forums.spoken-tutorial.org>
- Choose the minute and second where you have the question
- Explain your question briefly
- Someone from the FOSSEE team will answer them



# DWSIM Flowsheeting Project

- The FOSSEE team coordinates conversion of existing flow sheets
- We give honorarium and certificates for those who do this
- For more details, please visit this site  
<http://dwsim.fossee.in/flowsheeting-project>



# Textbook Companion Project

- The FOSSEE team coordinates coding of solved examples of popular books
- We give honorarium and certificates for those who do this
- For more details, please visit this site  
<http://dwsim.fossee.in/textbook-companion-project>





# Lab Migration Project

- The FOSSEE team helps migrate commercial simulator labs to DWSIM
- We give honorarium and certificates for those who do this
- For more details, please visit this site  
<http://dwsim.fossee.in/lab-migration-project>



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# Thanks

- Thanks for joining

