

# Plug Flow Reactor

Spoken Tutorial Project  
<http://spoken-tutorial.org>

National Mission on Education through ICT  
<http://sakshat.ac.in>

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# Learning Objectives

**In this tutorial, we will learn:**



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In this tutorial, we will learn:

- Define a **Kinetic reaction**



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- Simulate a **Plug Flow Reactor(PFR)**



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In this tutorial, we will learn:

- Define a **Kinetic reaction**
- Simulate a **Plug Flow Reactor(PFR)**
- Calculate **Conversion and Residence time** for a reaction in a PFR



# System Requirement



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- DWSIM v 4.3



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- Windows 7



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- DWSIM v 4.3
- Windows 7
- Any OS: Linux, Mac OS X or FOSSEE OS on ARM.



# Prerequisites

To practice this tutorial, you should know



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- How to add components to a flowsheet



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To practice this tutorial, you should know

- How to add components to a flowsheet
- How to select thermodynamic packages
- How to add material and energy streams and specify their properties



# Prerequisite Tutorials and Files

- <http://spoken-tutorial.org>
- You can access these tutorials and all the associated files from this site



# Reaction, Package and Inlet Condition

<b>Reaction</b>	$\text{N}_{2(g)} + 3\text{H}_{2(g)} \rightleftharpoons 2\text{NH}_{3(g)}$	
<b>Package</b>	<b>Peng Robinson (PR)</b>	
<b>Inlet Stream</b>	<b>Mass Flow</b>	<b>3600 kg/h</b>
	<b>Temperature</b>	<b>425 °C</b>
	<b>Pressure</b>	<b>200 bar</b>
	<b>Mole fraction</b>	$x_{\text{N}_2} = 0.5$ $x_{\text{H}_2} = 0.5$



# Reactor Parameters and Reaction Kinetics

<b>Reactor type</b>	<b>Isothermal</b>
<b>Reactor Parameters</b>	<b>Volume= 1 m<sup>3</sup> Length= 1.5 m</b>
<b>Reaction Kinetics</b>	<b><math>r_A = KC_a^n</math> <math>K = 0.004</math> <math>n = 1</math></b>



# Summary

In this tutorial, we have learnt to:

- Define a Kinetic reaction
- Simulate a Plug Flow Reactor (PFR)
- Calculate Conversion and Residence time for a reaction



# Assignment

Repeat the simulation with

- **Different compounds**

Ethylene oxide, Water & Ethylene glycol



- **Different thermodynamics**

Raoult's Law



# Assignment

Repeat the simulation with

- **Different feed conditions**

Mass Flow: 1000 kg/h

Mole Fraction( $\text{C}_2\text{H}_4\text{O}$ ): 0.2

Mole Fraction( $\text{H}_2\text{O}$ ): 0.8

Temperature: 55 °C

Pressure: 1 bar



# Assignment

Repeat the simulation with

- Different PFR dimensions

Volume: 1 m<sup>3</sup>

Length: 1.2 m

- Different reaction kinetics

$$r_A = KC_a^n, K = 0.005 \text{ 1/s}, n = 1$$



# About the Spoken Tutorial Project

- Watch the video available at [http://spoken-tutorial.org/What\\_is\\_a\\_Spoken\\_Tutorial](http://spoken-tutorial.org/What_is_a_Spoken_Tutorial)
- It summarises the Spoken Tutorial project



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- It summarises the Spoken Tutorial project
- If you do not have good bandwidth, you can download and watch it



# Spoken Tutorial Workshops

## The Spoken Tutorial Project Team,

- Conducts workshops using spoken tutorials
- Gives certificates to those who pass an online test
- For more details, please write to [contact@spoken-tutorial.org](mailto:contact@spoken-tutorial.org)



# Forum for specific questions

- Do you have questions in this Spoken Tutorial?
- Please visit <http://forums.spoken-tutorial.org>
- Choose the minute and second where you have the question
- Explain your question briefly
- Someone from the FOSSEE team will answer them



# Textbook Companion Project

- The FOSSEE team coordinates coding of solved examples of popular books
- We give honorarium and certificates for those who do this
- For more details, please visit this site <http://dwsim.fossee.in/textbook-companion-project>



# Lab Migration Project

- The FOSSEE team helps migrate commercial simulator labs to DWSIM
- We give honorarium and certificates for those who do this
- For more details, please visit this site <http://dwsim.fossee.in/lab-migration-project>



# DWSIM Flowsheeting Project

- The FOSSEE team coordinates conversion of existing flow sheets
- We give honorarium and certificates for those who do this
- For more details, please visit this site <http://dwsim.fossee.in/flowsheeting-project>



# Acknowledgements

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# Thanks

- Thanks for joining

